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ESTIMATION OF TOTAL IRON AND TRANSFERRIN BOUND IRON (TBI) IN HUMAN SERUM USING ICP-OES AND ITS APPLICATION ON BIOEQUIVALENCE STUDIES OF IRON-SUCROSE INJECTION

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INTRODUCTION

1.6 Billion people are affected by Iron-deficiency anemia worldwide. IV Iron therapy commonly administered to dialysis patients as an adjunct to managing anemia provides a potentially rich source for intradulyte bioactive from Iron-Sucrose, Iron-decturan or ferric glecomide complex like vender or Inferring substantially the concentration of the Iron-Sucrose, Iron-decturan or ferric glecomide complex like vender or Inferring substantially increased. How deficiency primarily in patients on chronic disabilities also in patients unable to take oral iron supplements. After parenteral administration, the concentration of total from remain substantially increased. Most colorimetric analysis of Iron detection primarily in patients on the Iron-Sucrose in the Iron-Sucrose in Iro against parameters of the control of results in overestimation of TBI present in serum. An inaccurate measurement of TBI may cause false detection of TBI over saturation indicating existence of free Iron, a condition that may cause

A need therefore exists for a method & equipment for isolation of Iron components in serum that contain Iron-Sucrose complex. Further, We also need a method for isolating and directly measuring amount of TBI and the amount of Iron carbohydrate present in the serum after administration of Iron-sucrose complex. Pharmacokinetic studies of IV Iron sucrose are complicated by backgrocirculating iron levels as well as need to differentiate and independently monitor Iron-sucrose and TBI. Prior to administration of product, most circulating from is in the form of TBI. Following therapy, circulating Iron is both as component of dosed formulation & complexed to transferrin. The typical colorimetric assays are incapable of discriminating and accurately quantifying TBI or total iron levels following iron sucrose administration. The measurement of TBI requires separating & excluding free Iron and drug derived iron from TBI.

We have developed and validated a robust and accurate method for analysis of total Iron and TBI to generate Pharmacokinetic data as component of Bioequivalence studies for generic IV. Th isurement device used for total Iron and Transferrin Bound Iron was ICP-OES iron from Iron sucrose complex. Circulating Iron-Sucrose could then be calculated as difference between total

MATERIALS

Iron solution, IG/dl used as Standard was purchased from Fisher scientific while Indium Standard, IG/dl purchased from Fluka.

Sodium Nitrate Sodium Chloride purchased from Merck Cysteine Hydrochloride Monohydrate purchased from Lobachemicals and Calcium Nitrate Tetrahydrat

Holo transferrin Human purchased from sigma was used for preparation of TBI QCs. Nitric acid purchased from Merck while HPLC water purchased from Rankern. Alumina Basic Cartridges, 500mg/3 ml. Extraction cartridges purchased from Orochem.



EOUIPMENT

A Thermo Scientific Inductive Couple Plasma System Equipped with Optical Emission Spectrophotometer (7000 Series) and System controlled through Qtegra Software

TOT ONE COMMITTEE	View Direction	Axial
	UV Exposure Time	15
	UV RF Power	1150
	UV Neb Gas Flow	0.5
	VIS Exposure Time	5
	VIS RF Power	1150
	VIS Nebulizer Gas Flow	0.5



METHOD VALIDATION

Method Used for Determination of Iron and Transferrin Bound Iron in Human plasma was developed and Validated for ICP-OFS

stability performed at HQC and LQC level in Human serum and Simulated Serum a

Method Validated in terms of System Suitability, Linearity, Accuracy & Precision, Extraction Efficiency, Dilution Integrity and Stability experiment.

Holo Transferrin Human was added into Human Serum. system suitability performed by aspirating 6 replicate of higher standard. Calibration standard used for linearity were prepared in Simulated serum.

QCs were used for Accuracy & Precision determination were prepared in I stion of HOC MOC and LOC leve Dilution Integrity evaluated for dilution Factor 2 and 5 times of High QC. itock Solution Stability Performed at HQC and LQC level at -20±5 °C reeze thaw stability performed at HQC and LQC level in Human serum at -20±5 °C

TBI





RESULTS

NOVEL ASPECTS

Since colorimetric assays rely on acidic release of Iron from Transferrin, they are not very accurate as Iron-Carbohydrate complex is very stable in acidic conditions and they are also not capable of

At Intervein, we developed and validated a robust and accurate method for analysis of Total Iron and Transferrin Bound Iron to generate pharmacokinetic data as a component of bioequivalence studies intravenous preparation. We have developed accurate method to separate TBI using alumina basic columns and the final concentrations were calculated by subtracting TBI Conc from Total Iron Concentration

We have also compared the results of colorimetric assay and estimation by ICP-OES. The results is as below:

115.10

138.80

32.00

27.00

35.20

 (µg/dL)
 (µg/dL)
 (µg/dL)
 (µg/dL)
 (µg/dL)

 32.00
 119.54
 12.32
 76.60

12.32

47.20

9.12

9.56

	-							10	6014.60	21513.40	340.75	12137.50		
4	_	69.13	260.00	14.88	80.30		2	15	4188.10	10021.60	157.31	5848.90		
5		27.87	116.60	8.53	45.60			20	6301.00	7169.40	112.10	3880.60		
The res	The result of test and reference drug for total iron and Transferrin Bound Iron is as below:													
20 20 20 20 20 20 20 20 20 20 20 20 20 2	Linear Fiel Since						990.0 179	2 200 4.00	Lb	No Made Nado Nado	26:00 22:00 24:00 24	7 . G. PARAMONTAL		
Mean Linear Plot of Iron For Test & Reference Product Mean Linear Plot of TB1 For Test & Reference Product														

REFERENCES

nical Society Publications: Anal. Chem. 1989, 61 (17), pp 1851-185 FDA Guideline: Recommended Mar 2012; Revised Nov 2013 16, pages 1837-1846, DOI 10, 4155/bio.11.180

L Guideline for industry-Biographtical Method Validation, May-2001 "Labile Iron in parenteral iron formulations: a quantitative (NO4) 19: 561-565

4740.90 18203.10

340.75

157.31

112.10

2354.50

3109.40

3395.50

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